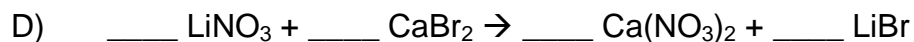
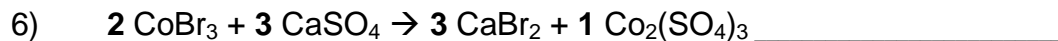
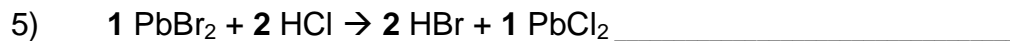
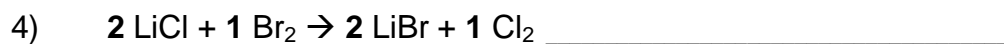
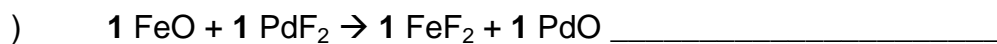


Exam Revision

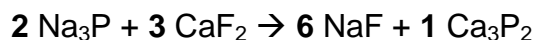
Balance the following equations

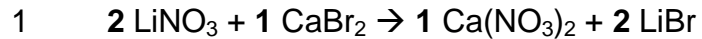


Classify the following equations:



Calculation questions:

7) How much 0.89 M of Na_3P is required in (mL) to react with 0.89 g of CaF_2 . Show all of your working out.8) If 0.89 g of LiO_3 and 2.89 g of CaBr_2 are used, determine which is the limiting reagent to produce LiBr . Show all working out.



9. Determine what the definition of a titration is :

10. Define equivalence point

11. Reflection on some of the issues that were discussed in class:

1. Is science always beneficial or always a hindrance

2. Has society always benefited from science invention?

3. How has technology changed or improved science discoveries?